

Top-down patterning and self-assembly of flower-like gold arrays for surface enhanced Raman spectroscopy



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ABSTRACT

A low-cost and controllable fabrication method of flower-like Au arrays has been developed by using photosensitive sol–gel and electrochemical reaction self-assembly strategy on a TiO₂ patterned Si substrate. The effects of concentration of HAuCl₄ solution and applied voltage have been investigated to tailor the growth of Au nanoflowers. At a proper applied voltage (12 V) and concentration (0.005 mM), fractal flower-like Au arrays is prepared and shows strong SERS effect. The concentration of detectable Rhodamine 6G can be down to 10⁻¹⁰ M using the short data integration time (5 s). These flower-like Au arrays with strong SERS effect could be used in biosensors and nanodevices with molecule-level detection.

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1. Introduction

As a powerful and extremely sensitive spectroscopic technique, Surface-enhanced Raman Scattering (SERS) has attracted much attention in the past decade due to its wide applications in analytical chemistry, biomedical detection, and environmental monitoring [1–4]. For practical application, the ideal SERS substrates should have sufficiently high enhancement factors, and also be uniformity, reproducibility and facile synthesis. Noble metals such as gold (Au), silver (Ag) and copper (Cu) nanostructures should be the ideal SERS substrates for their localized surface plasmon resonance (LSPR), which show strong SERS enhancement [5–9]. In particular, due to the rapid development of micro or nano lithography, various ordered noble metallic arrays have been designed and engineered to work as the SERS substrates. Compared to smooth and random SERS substrates, these ordered metallic arrays could generate high electric field enhancement, and achieve broadband and unidirectional excitation of SPPs, even at normal incidence [10–12]. However, ordered metallic arrays based bottom-up approach face significant challenge of producing reproducible and uniform arrays [13], while, top-down approach such as electron or ion beam lithography could produce more uniform and reproducible nanostructures. But these

techniques not only require time consuming serial processing in vacuum, but also are limited and very costly due to the requirements of specialized equipment with controlled environments [14,15]. Nanoimprint lithography could produce nanostructures with the advantages of lower cost, ease of use than the capital intensive equipment, however, still it may require the use of an elastomeric mold patterned with relief structures on its surface and after curing, the mold should be peeled away carefully [16,17].

In the work reported here, we combined top-down and bottom-up approaches, developing a low-cost and controllable method for flower-like Au arrays by means self-assembly and electrochemical reaction occurring inside the confined region by patterned organic–inorganic TiO₂ film microstructure on Si substrates. Specifically, the patterned TiO₂ microstructure was prepared by photosensitive sol–gel, which make the TiO₂ films with photosensitive properties through chelating the metal alkoxide titanium tetrabutoxide monomer (Ti(O(CH₂)₃CH₃)₄: Ti(O-n-Bu)₄) with the chemical modifier benzoylacetone (C₆H₅COCH₂COCH₃: BzAcH). The major advantage of this approach is that you can obtain inorganic or organic–inorganic composited materials microstructures even nanostructures with different sizes simply and rapidly by photo lithography but not using photoresist as sensitive medium [18–20]. At the same time, the flower-like Au arrays prepared by self-assembly and electrochemical reaction can grow tightly on the substrate and importantly show strong SERS effect. The details are reported in this paper.

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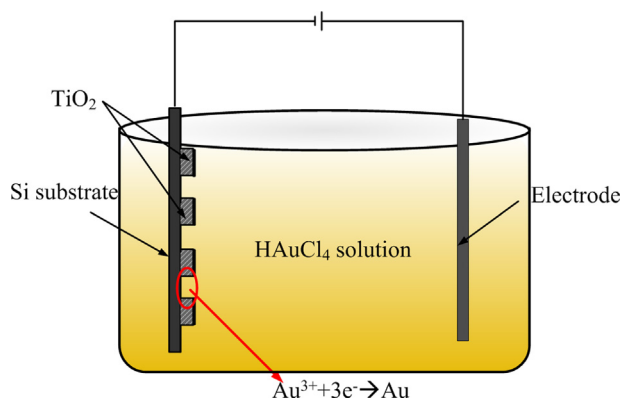


Fig. 1. Schematic illustration of the electrochemical system for fabricating the flower-like Au arrays.

2. Experimental

The TiO₂ photosensitive sol is synthesized by refluxing the Ti(O-n-Bu)₄ and BzAcH in the absolute methanol (CH₃OH:MeOH) with the molar ratio is 1:1:40 for 2–3 h. The homogeneous photosensitive TiO₂ gel films are prepared on Si substrates by dip-coating. And then, TiO₂ microstructure are fabricated by irradiating the gel film with UV light (Ushio SP-9 250DB, power 250W) through a mask followed by dissolving the non-irradiated area in a suitable solvent. After that, the microstructure should be heat-treated at 250 °C for 30 min to make it stable and reliable.

For a typical synthesis of flower-like Au arrays, the electrolyte containing HAuCl₄ solution is first prepared with distilled water. The patterned organic–inorganic TiO₂ microstructure on Si substrate is used as working electrode (or cathode), and a clean Zn sheet is used as auxiliary electrode (see Fig. 1). Because the organic–inorganic TiO₂ is insulated, the electrochemical reaction will occur mostly at the bare Si substrate. Thus, flower-like Au

arrays can be generated. The effects of concentration of HAuCl₄ solution and applied voltage have been investigated to tailor the growth of Au nanoflowers. The sample is cleaned with distilled water for several times and dried before microstructural characterization and SERS measurement.

SERS measurements are carried out using a HORIBA Jobin Yvon LabRAM HR Evolution confocal microprobe Raman spectrometer with the 532 nm He–Ne Laser line at room temperature. Samples for SERS are dipped into the Rhodamine 6G (R6G) aqueous solution with concentrations of 10^{−5} M for 10 h, rinsed with de-ionized water, and dried before measurement.

3. Results and discussion

Fig. 2 shows the SEM images of the as-prepared flower-like Au arrays on Si substrates. As shown in Fig. 2, the dark-colored part is TiO₂ film, while light-colored part is bare Si substrate. Owing to the electrical insulativity of organic–inorganic TiO₂, the electrochemical reaction occur only at the bared Si substrate, at where [Au(OH)₄][−] has been reduced to flower-like Au(solid) [21]. Thus, an array of flower-like Au is generated on patterned Si substrate. The average diameter of each Au nanoflowers is about 60–120 nm.

Preferential growth of Au nanoflowers on TiO₂ patterned Si substrate has been investigated by changing the concentration of HAuCl₄ solution and applied voltage. Fig. 3 shows the SEM images of the Au nanoflowers on TiO₂ patterned Si substrate prepared with the applied voltage of 28 V (Fig. 3(a)), 24 V (Fig. 3(b)), 20 V (Fig. 3(c)), 16 V (Fig. 3(d)), 12 V (Fig. 3(e)), 8 V (Fig. 3(f)), 4 V (Fig. 3(g)) and 2 V (Fig. 3(h)), respectively. The experiments results show that: when the voltage is higher than 16 V, the isolated island Au nanostructure is formed on Si substrate. In contrast, reducing the applied voltage obviously facilitates the fractal growth of the Au nanoflowers, and meanwhile, the density of Au nanoflowers will be reduced. When the voltage is quite low (less than 2 V), the Au nanoflowers would not grow.

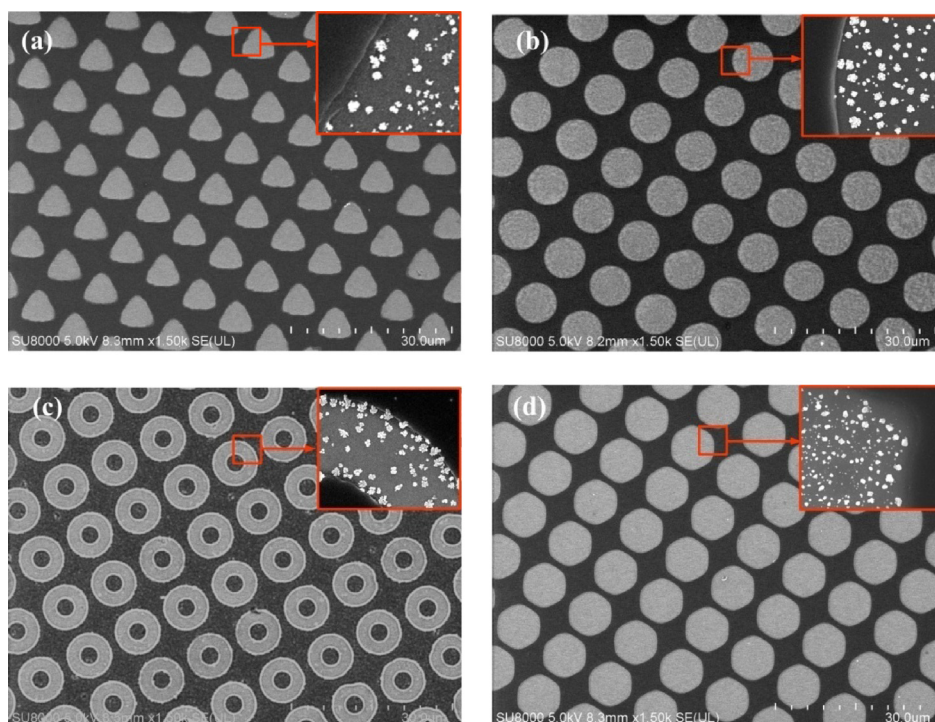


Fig. 2. SEM images of the as-prepared flower-like Au arrays using different TiO₂ microstructure on Si substrates. The concentration of HAuCl₄ solution is 0.005 mM.

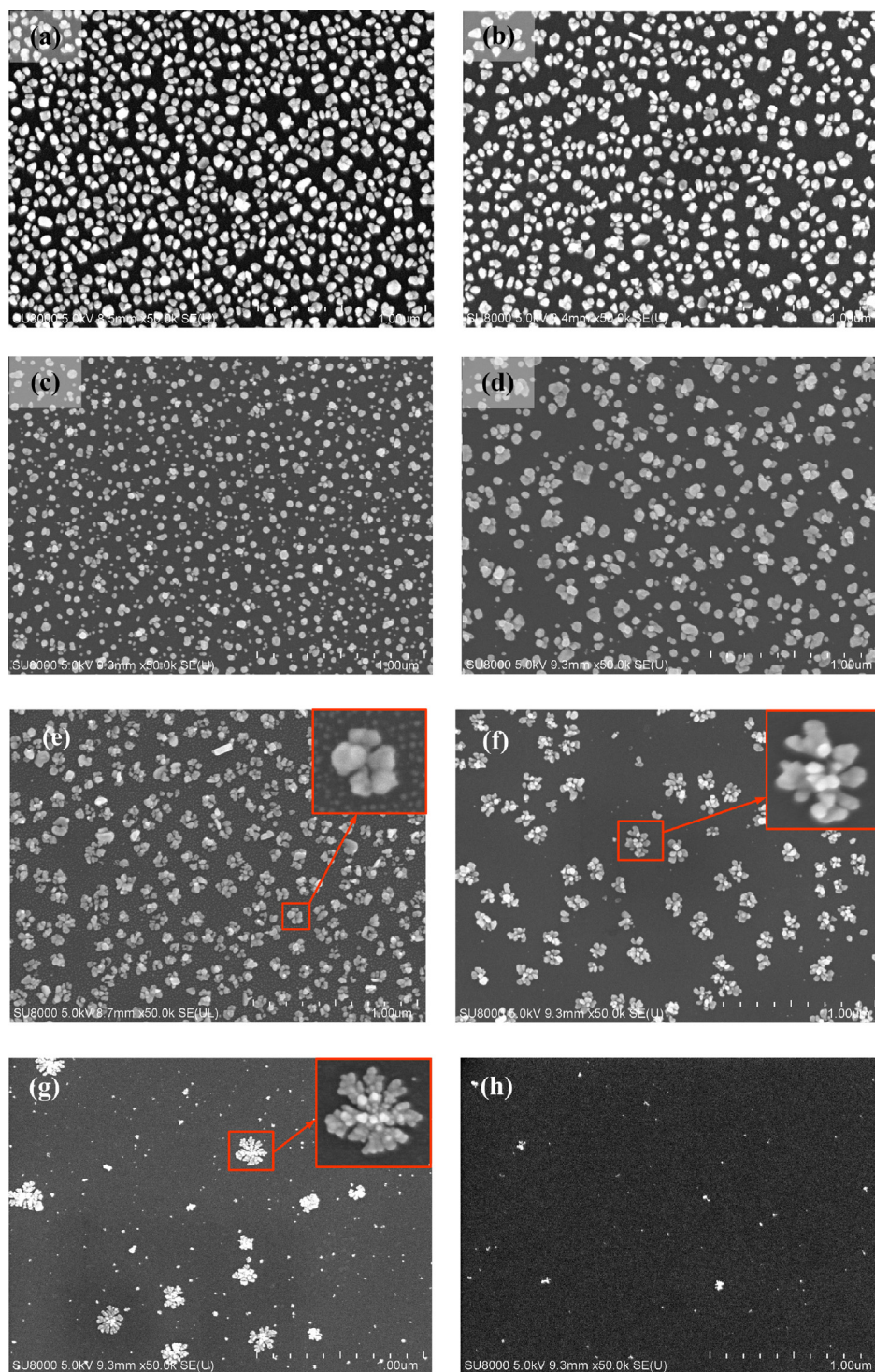


Fig. 3. SEM images of the Au nanoflowers on TiO_2 patterned Si substrate prepared with different applied voltage. The voltage of (a), (b), (c), (d), (e), (f), (g) and (h) is 28 V, 24 V, 20 V, 16 V, 12 V, 8 V, 4 V, 2 V, respectively. The concentration of HAuCl_4 solution is 0.005 mM and the deposition time is 40 min.

The concentration of HAuCl_4 solution has been played another important role in deciding which Au architecture is formed. Fig. 4 shows the SEM images of the Au nanoflowers on TiO_2 patterned Si substrate prepared with different concentration of HAuCl_4 solution. The concentration of (a), (b), (c) and (d) is 0.2 mM, 0.1 mM, 0.01 mM and 0.005 mM, respectively. From the experiment, we find that when concentration of HAuCl_4 solution is higher, isolated island or spherical Au nanostructure is formed as shown in Fig. 4(a)–(c). Decreasing concentration to 0.001 mM, very little Au

can be deposited onto the Si substrate. Keeping the concentration of HAuCl_4 solution is 0.005 mM, fractal Au nanoflowers is easily obtained (see Fig. 4(d)).

On the basis of the previous discussion, the formation mechanism of Au nanoflowers can be illustrated by the schematic drawing in Fig. 5. Changing the concentration of HAuCl_4 solution and applied voltage gives one the ability to control the nucleation and growth processes through careful regulation of reaction rate. At elevated concentration or applied voltage, the

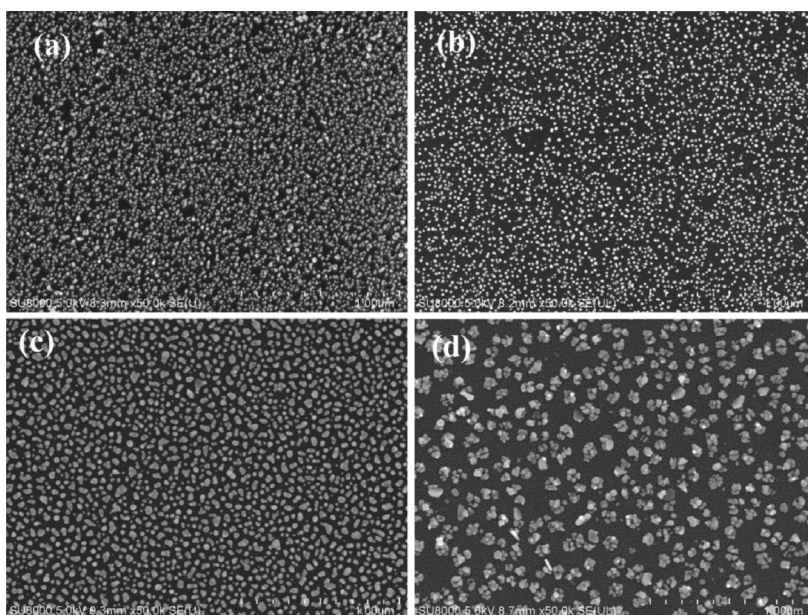


Fig. 4. SEM images of the Au nanoflowers on TiO₂ patterned Si substrate prepared with different concentration of H[AuCl₄] solution. The concentration of (a), (b), (c) and (d) is 0.2 mM, 0.1 mM, 0.01 mM and 0.005 mM, respectively. The applied voltage is 12 V and the deposition time is 40 min.

reaction rate of Au³⁺ ions into Au atoms can be accelerated and thereby to accelerate the nucleation and isotropic growth steps (see Fig. 5(a)). The selected-area electron diffraction (SAED) pattern indicates that the spherical Au nanostructure exhibits

single crystal structure. In contrast, lower concentration or applied voltage can reduce the reaction rate of Au³⁺ ions into Au atoms, which facilitates the anisotropic growth and formation of fractal Au nanoflowers (see Fig. 5(b)). The SAED pattern indicates

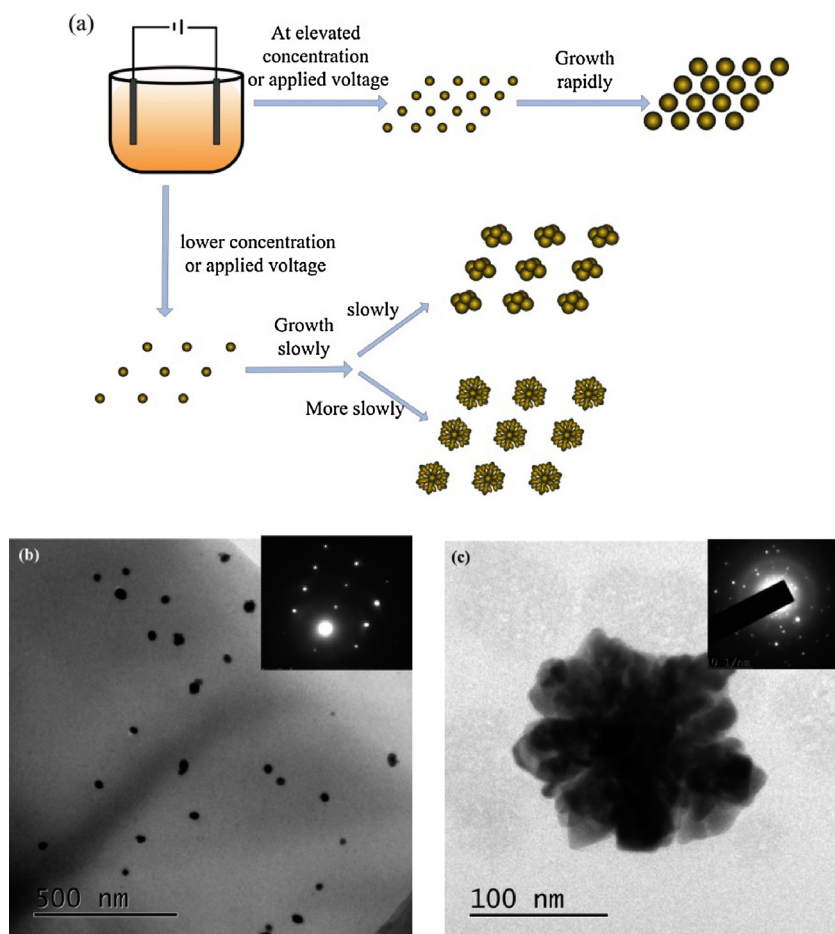


Fig. 5. Schematic drawing of the formation mechanism of Au nanoflowers. (b) and (c) is the TEM image of spherical Au nanostructure and fractal Au nanoflowers, respectively.

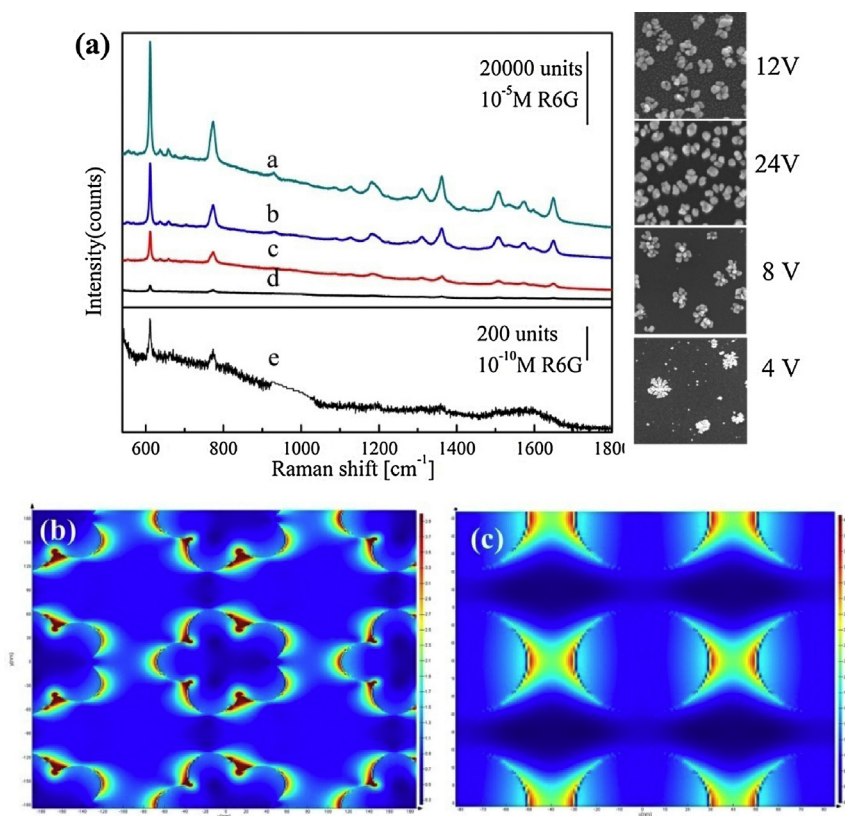


Fig. 6. SERS spectra for as-prepared flower-like Au array probed with (data integration time is 5 s). Curves a–d in (a) are SERS spectra probed with 10^{-5} M R6G corresponding to the Au arrays shown in Fig. 3(b), (a), (c) and (d). Curve e is SERS spectrum for flower-like Au array probed with 10^{-10} M R6G (data integration time: 5 s). (b) The spatial distribution of the electric field obtained by 3D FDTD simulation for flower-like Au array with the applied voltage of 12 V. (c) The spatial distribution of the electric field obtained by 3D FDTD simulation for spherical Au array with the applied voltage of 24 V.

that the structure of Au nanoflowers is polycrystalline, which can be ascribed to the random growth and aggregation model of the nanocrystals [22]. The detail growth mechanism of Au nanoflowers would be investigated and discussed in our further research.

The as-grown flower-like Au array show strong SERS effect by using R6G as probe molecules (see curves in Fig. 6(a) corresponding to different applied voltages). The SERS signal from Au nanoflowers is enhanced much more than from the isolated island Au nanostructure (see curves a and b of Fig. 6). In addition, from the curves c and d, it can be seen that the flower-like Au array with less Au nanoflowers (applied voltage: 8 V and 4 V) is of weaker SERS signal. Furthermore, in our experimental, the SERS signal has been obviously detected with 10^{-10} M R6G using the same data integration time (5 s) as seen in Fig. 6 curve e, which shows the possibility of molecule-level detection. The spatial distribution of electric field of Au array with the flower-like nanostructure and spherical nanostructure is studied by three dimensional finite-difference time-domain (3D FDTD) simulations (Lumerical FDTD Solution, Inc., Canada) to make a direct comparison with the experimental SERS results (see Fig. 6(b) and (c)) [23]. The total field/scattered-field plane wave source with the wavelength of 532 nm is selected and the polarization direction of is parallel to the horizontal direction. The mesh size of the override mesh region is set to be 1 nm. As shown in Fig. 6(b) and (c), the electromagnetic field (EM) enhancement in the gaps generated from Au array with the flower-like nanostructure is larger than the spherical-like nanostructure. These gaps are vital hot spots for SERS, which

indicate that the enhancement factor of flower-like Au array is stronger than spherical-like Au array. The FDTD simulation results are consistent with the SERS spectra.

4. Conclusions

In summary, cost-effective SERS substrates with ordered Au arrays are successfully fabricated by means of photosensitive sol-gel and electrochemical reaction self-assembly on TiO₂ patterned silicon slides. The micro morphology of Au arrays can be controlled and tailored by modifying the concentration of electrolyte solution and applied voltage simply. In this study, fractal flower-like Au arrays prepared at the applied voltage of 12 V and concentration of 0.005 mM shows very strong SERS effect. The present investigation opens a possibility of applying such ordered fractal metallic arrays as SERS substrates offering excellent electric field enhancement and further interesting properties such as broadband and excitation of SPPs, which is of great importance for be used in biosensors and nanodevices with molecule-level detection.

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